

Alkaloid Synthesis

Deutsche Ausgabe: DOI: 10.1002/ange.201505604 Internationale Ausgabe: DOI: 10.1002/anie.201505604

Synthesis of $\it ent$ -Ketorfanol via a C–H Alkenylation/Torquoselective 6π Electrocyclization Cascade

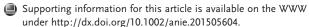
Eric M. Phillips, Tehetena Mesganaw, Ashay Patel, Simon Duttwyler, Brandon Q. Mercado, Kendall N. Houk,* and Jonathan A. Ellman*

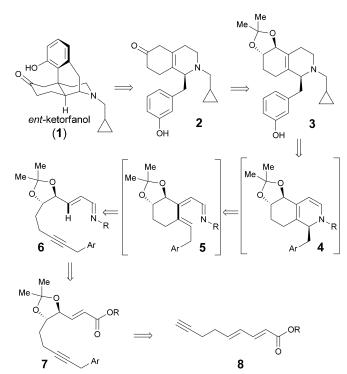
Abstract: The asymmetric synthesis of ent-ketorfanol from simple and commercially available precursors is reported. A Rh^{l} -catalyzed intramolecular C-H alkenylation/torquoselective 6π electrocyclization cascade provides a fused bicyclic 1,2-dihydropyridine as a key intermediate. Computational studies were performed to understand the high torquoselectivity of the key 6π electrocyclization. The computational results demonstrate that a conformational effect is responsible for the observed selectivity. The ketone functionality and final ring are introduced in a single step by a redox-neutral acid-catalyzed rearrangement of a vicinal diol to give the requisite carbonyl, followed by intramolecular Friedel-Crafts alkylation.

Oxycodone^[1] and related semisynthetic opioid ligands are effective and pervasively used medications for the management of pain. However, these compounds have significant liabilities that include dependency, tolerance, which results in increasing doses being required to maintain efficacy, and the serious side effects of higher doses, such as respiratory failure. The recently reported high-resolution structures of a number of opioid receptor ligands in complex with several receptor subtypes provide an exceptional opportunity for the structure-based design of new opioid ligands with reduced drawbacks.^[2] However, access to different structural variants is constrained by reported synthetic routes to the approved semisynthetic opioid ligands, which rely on morphine, thebaine, and other morphinoid natural products as heavily functionalized starting materials.^[3,4]

Our recently developed Rh-catalyzed C–H functionalization/ 6π electrocyclization cascade^[5] should enable direct and rapid entry to this important class of alkaloids from readily available achiral precursors. Herein, we demonstrate this approach with the asymmetric synthesis of *ent*-ketorfanol (1), the nonregulated enantiomer of the semisynthetic opioid drug ketorfanol (Scheme 1). *ent*-Ketorfanol should be accessible through acid-catalyzed intramolecular Friedel–Crafts cycliza-

[*] E. M. Phillips, T. Mesganaw, S. Duttwyler, B. Q. Mercado, Prof. Dr. J. A. Ellman
 Department of Chemistry, Yale University
 225 Prospect St., New Haven, CT 06520 (USA)
 E-mail: jonathan.ellman@yale.edu
 A. Patel, Prof. Dr. K. N. Houk
 Department of Chemistry and Biochemistry
 University of California, Los Angeles
 Los Angeles, CA 90095-1569 (USA)
 E-mail: houk@chem.ucla.edu





Scheme 1. Retrosynthetic analysis of ent-Ketorfanol.

tion of intermediate 2.^[6] This intermediate could potentially be obtained in situ from 3 under the acidic reaction conditions via a speculative redox-neutral process^[7] proceeding through ionization of the allylic oxygen functionality followed by a hydrogen shift (see below). Fused bicyclic tetrahydropyridine 3 would then be obtained through reduction of dihydropyridine 4 under mildly acidic reductive amination conditions. In the key step, bicyclic intermediate 4 should be accessible via an intramolecular Rh-catalyzed syn C-H bond addition to the alkyne, followed by electrocyclization, which should hopefully proceed with high torquoselectivity owing to remote asymmetric induction provided by the isopropylidine protected diol. [8-10] The precursor for this cascade reaction, imine 6, should be readily accessible from ester 7 through straightforward functional-group transformations. Ester 7 should be obtainable by regioselective Sharpless asymmetric dihydroxylation,[11] isopropylidene protection, and alkyne benzylation of the achiral dienoate 8.

The synthesis commenced with Swern oxidation of 4-pentyn-1-ol (9) and subsequent Horner-Wadsworth-Emmons reaction (Scheme 2) to furnish dienoate 10 (60% yield, 2 steps). A highly regio- and enantioselective Sharpless

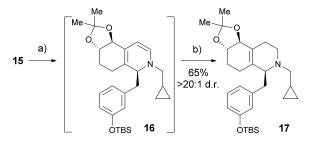


Scheme 2. Synthesis of imine **15**. Reagents and conditions: a) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, $-78\,^{\circ}$ C; b) nBuLi, iPr_2NH , (E)-ethyl-4-(diethoxyphosphoryl)but-2-enoate, $-78\rightarrow 0\,^{\circ}$ C; c) AD-mix-α, MeSO₂NH₂, tBuOH, H₂O, 0 $^{\circ}$ C; d) 2,2-dimethoxypropane, pTsOH (10 mol%), CH₂Cl₂, 0 $^{\circ}$ C; e) 3-TBSOPhCH₂Cl, PdCl₂(CH₃CN)₂ (5 mol%), XPhos (15 mol%), Cs₂CO₃, THF, 65 $^{\circ}$ C; f) DIBAL, CH₂Cl₂, $-78\,^{\circ}$ C; g) Dess–Martin periodinane, pyridine, CH₂Cl₂, $0\,^{\circ}$ C; h) cyclopropylmethyl amine, toluene, 3 Å MS. DMSO = dimethyl sulfoxide, TsOH = 4-toluenesulfonic acid, TBS = tert-butyldimethylsilyl, THF = tetrahydrofuran, DIBAL = diisobutylaluminium.

asymmetric dihydroxylation was performed on the olefin distal to the ester (81% yield, 95% ee). Diol **11** was immediately protected using 2,2-dimethoxypropane and pTsOH to furnish acetonide **12** in 94% yield. X-ray structural analysis of **11** was also carried out to confirm that its absolute configuration was consistent with that predicted for AD-mix- α . Discrepance of the distribution of the dist

Alkyne **12** was next benzylated by Cu^I -mediated coupling to give a moderate yield of product **13** (50–70%). [13] However, this process was plagued by long reaction times (4 days), the need for stoichiometric Cu^I , and difficulty in obtaining pure material. Ultimately, the efficient Pd-catalyzed method developed by Buchwald and co-workers [14] enabled the coupling of benzyl chloride and terminal alkyne **12** in 75% yield (Scheme 2). DIBAL reduction and Dess–Martin periodinane oxidation afforded α,β -unsaturated aldehyde **14** (54% yield, 2 steps). Condensation with cyclopropylmethyl amine gave our key imine substrate **15** in 95% yield.

With α,β -unsaturated imine **15** in hand, we attempted the crucial tandem Rh^I-catalyzed intramolecular alkenylation/ 6π electrocyclization (Scheme 3).^[5] Notably, while previously reported examples of diastereoselective electrocyclizations of azatrienes rely on stereogenic centers directly attached to the two atoms involved in ring closure,^[8-10] azatriene **5** incorpo-



Scheme 3. The Rh¹-catalyzed C—H insertion reaction. Reagents and conditions: a) [{RhCl(coe) $_2$ } $_2$] (2.5 mol%), 4-(diethylphosphino)-*N*,*N*-dimethylaniline (5 mol%), toluene, 65 °C; b) NaHB(OAc) $_3$, AcOH, EtOH, $0 \rightarrow 23$ °C.

rates vicinal diol stereocenters that are distant from the stereocenter to be formed. Exposure of imine **15** to 2.5 mol % of [{RhCl(coe)₂}₂] and 5 mol % of the electron-rich ligand 4-(diethylphosphino)-*N*,*N*-dimethylaniline at 65 °C led to highly selective formation of dihydropyridine **16** in excellent yield (84 % by ¹H NMR analysis). While this tandem reaction was found not to be overly sensitive to concentration, reaction temperature is important, with higher temperatures resulting in lower yields and the formation of isomeric byproducts. Intermediate **16** was reduced without isolation by using NaHB(OAc)₃ and AcOH to provide tetrahydropyridine **17** in 65 % overall yield and greater than 20:1 d.r., thereby confirming the high torquoselectivity of the reaction.

To determine the origins of the torquoselectivity observed in the Rh^{I} -catalyzed cascade sequence, computional studies of the key aza-electrocyclization were conducted using the $\omega B97x$ -D density functional with the 6-31+G(d,p) basis set. Table 1 illustrates the truncated substrates modeled computationally as well as the kinetic diastereoselectivities of ring closures of these compounds. The sense and level of the torquoselectivities computed are consistent with the experimental results. The discussion of the stereoinduction (see below) emphasizes the results of our computational studies on E, the 1-azatriene that most closely resembles the experimental substrate E.

Table 1: 1-Aza-3,5-trienes modeled computationally, computed activation free energy differences, and computed torquoselectivities.

Compound	R ¹	R ²	R^3	G^{\dagger} [kcal mol ⁻¹]	Torquoselectivity ^[a]
A	Н	Н	Н	3.1	187:1
В	Н	Me	Н	3.3	262:1
C	Me	Me	Н	3.6	435:1
D	Н	Me	Me	3.1	187:1
E	Me	Me	Me	3.2	221:1

[a] Free energies and torquoselectivities determined assuming a standard state of 1 atm and 298.15 K.



The torquoselectivity of the electrocyclization is controlled by the transition-state conformational preferences of the six-membered ring appended to the 1-azatriene. In the favored transition structure **c-tsE**, the six-membered ring adopts a distorted chair-like arrangement, whereas this ring is forced into a twist boat arrangement in **tb-tsE**, thereby destabilizing this transition structure by 3.2 kcal mol⁻¹. Arguments first proposed to rationalize electrophilic additions to cyclohexenes and their corresponding oxides^[15] can be extended to rationalize these conformational differences. The favored mode of disrotation forces C5 of the triene moiety to pyramidalize "downwards" (as shown in Figure 1),

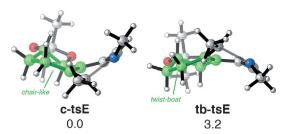


Figure 1. ω B97x-D/6-31 + G(d,p)-optimized transition structures and their relative free energies (in kcal mol⁻¹).

thus allowing the six-membered ring to adopt a chair-like conformation; pyramidalization of this carbon in the opposite direction is responsible for the twist-boat arrangement found in **tb-tsE**. This effect is likely general and could be exploited to perform diastereoselective electrocyclizations of similar chiral cyclic trienes.^[16]

Figure 2 depicts the free energy surface of both modes of disrotatory electrocyclization as well as the reactant conformers of **E**. The global minimum conformer of **E** is **c-E1**, which features a chair-like six-membered ring and an s-trans arrangement of the azatriene moiety. **tb-E1** has a relative energy of 4.1 kcal mol⁻¹ and leads to the disfavored transition structure. The helical arrangement of the azatriene in this structure forces the fused ring to adopt the less stable twist-boat conformation, thus demonstrating that the geometry required to form **tb-tsE** is inherently strained. The reactive conformers **c-E2** and **tb-E2** also differ in energy by approximately 4 kcal mol⁻¹, suggesting that the magnitude of this conformational effect is similar in the ground and transition states.

With the concise asymmetric synthesis of intermediate 17 accomplished, the stage was set for completing the synthesis of *ent*-ketorfanol (1). The remaining tasks included cleavage of the silyl ether and ketal protecting groups, intramolecular Friedel–Crafts alkylation, and transformation of the diol into the desired ketone moiety. All of these steps could theoretically be accomplished under acidic conditions. We therefore attempted to carry out this sequence in a single step by subjecting 17 to phosphoric acid at elevated temperature, which resulted in a 2:1 inseparable mixture of 18 and *ent*-ketorfanol (1), favoring 18 in a combined 69% yield (Scheme 4). Under these strongly acidic conditions, rapid cleavage of the ketal and silyl protecting group occurs along

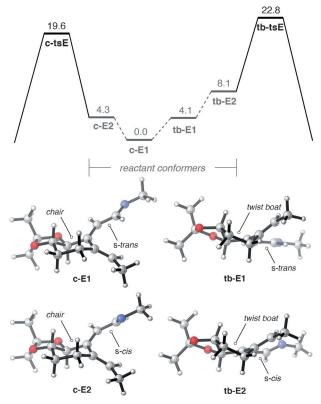
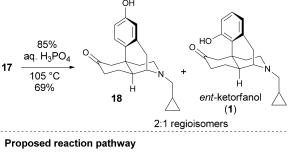


Figure 2. Free-energy diagram of the electrocyclization of 1-aza-3,5-triene **E** and the computed structures of the reactant conformers of **E**. Free energies (in kcal mol $^{-1}$) and structures were determined using ω B97x-D/6-31 + G(d,p).



Scheme 4. Intramolecular Friedel–Crafts alkylation with concomitant redox-neutral diol-to-ketone interconversion.

with ionization to give the stabilized 2°/3° allylic carbocation 19, which upon hydrogen migration affords ketone 20. This ketone intermediate accumulates as determined by NMR reaction monitoring. Subsequent intramolecular Friedel–Crafts alkylation of 20 then provides regioisomers 18 and 1.

The poor regioselectivity of the Friedel-Crafts reaction was not unexpected but did pose a significant challenge. An effective solution to this problem would be to block the



position para to the hydroxy group on the aromatic ring. However, a silicon blocking group would likely not survive the strongly acidic conditions necessary for the Friedel-Crafts cyclization and a bromine blocking group would likely be incompatible with the C-H functionalization conditions owing to competitive C-Br insertion.[17] We therefore targeted a substrate with a chloride para to the hydroxy group (Scheme 5). Benzyl chloride 21 was readily coupled with terminal alkyne 12 according to our previously described alkynylation conditions. Transformation of the α,β-unsaturated ester 22 into the required imine 24 then proceeded without incident.

We were now ready to subject imine 24 to our Rh^Icatalyzed C-H functionalization conditions (Scheme 6). At 55°C within 3 h, dihydropyridine 25 was produced in excellent yield as a single diastereomer (>95:5 d.r.). Subsequent reduction with NaHB(OAc)₃ provided tetrahydropyridine 26 in 69% yield of isolated product. Although the deactivating nature of the chloride substitution in 26 necessitated a higher temperature for the phosphoric acid mediated Friedel-Crafts alkylation compared to the des-chloro compound, cyclization to give 27 proceeded with excellent levels of regio- and diastereocontrol. Hydrodehalogenation of the aryl chloride in 27 with H₂ in the presence of Pd/C and NaHCO₃ cleanly afforded ent-ketorfanol (1). The structure and absolute configuration were rigorously established by X-ray crystallography.[12]

The asymmetric total synthesis of ent-ketorfanol was accomplished in 11 steps and 9% overall yield from commercially available achiral starting materials. In contrast, the only published routes to ketorfanol proceed through multistep degradations of morphine or naltrexone.[3] The RhI-catalyzed alkenylation/ 6π electrocyclization cascade rapidly furnishes a bicyclic 1,2-dihydropyridine that serves as a key intermediate in the synthesis. The electrocyclization is highly torquoselective and thus enables the use of a Sharpless asymmetric dihydroxylation to readily produce either enantiomer of

Scheme 5. Synthesis of imine 24. Reagents and conditions: a) PdCl₂-(CH₃CN)₂ (5 mol%), XPhos (15 mol%), Cs₂CO₃, THF, 65 °C; b) DIBAL, THF, -78 °C; c) Dess-Martin periodinane, pyridine, CH₂Cl₂, 0°C; d) cyclopropylmethyl amine, toluene, 3 Å MS.

Scheme 6. Synthesis of ent-ketorfanol (1). Reagents and conditions: a) $[\{RhCl(coe)_2\}_2]$ (5 mol%), 4-(diethylphosphino)-N,N-dimethylaniline (10 mol%), toluene, 55 °C; b) NaHB(OAc)₃, AcOH, EtOH, $0\rightarrow23$ °C; c) 85% phosphoric acid, 125°C; d) H_2 , Pd/C, $NaHCO_3$, EtOH.

ketorfanol. The acid-catalyzed tandem pinacol rearrangement/Friedel-Crafts alkylation not only provides the multicyclic drug framework but also converts the diol that served as the stereocontrolling element to the requisite ketone moiety in an efficient redox-neutral process.

In previous reports, we have demonstrated the versatility of our Rh^I-catalyzed C-H functionalization, electrocyclization, and reduction sequence for the regio- and stereoselective incorporation of diverse substituents into tetrahydropyridines.[5c-h] For this reason, the synthetic route reported herein should enable the straightforward preparation of a variety of ketorfanol analogues and thus provide an avenue for improving drug properties.

Acknowledgements

This work was supported by NIH Grant GM069559 (to J.A.E.). E.M.P. also acknowledges support from an NRSA postdoctoral fellowship (F32GM090661), and S.D. is grateful to the Swiss National Science Foundation for a postdoctoral fellowship (PBZHP2-130-966). We gratefully acknowledge Dr. Michael Takase for solving the crystal structure of 1·HCl. We also acknowledge the financial support of the NIH (GM-36700 and CHE-1351104 to K.N.H.) A.P. thanks the Chemical-Biology Interface Training Program for its support (T32 GM 008496). A.P acknowledges the University of California, Los Angeles for financial support. Density functional theory computations were performed using the Extreme Science and Engineering Discovery Environment (XSEDE)'s Gordon supercomputer (OCI-1053575) at the San Diego Supercomputing Center.

12215

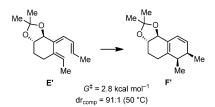


Keywords: alkaloids \cdot asymmetric synthesis \cdot C-H activation \cdot heterocycles \cdot torquoselectivity

How to cite: Angew. Chem. Int. Ed. **2015**, 54, 12044–12048 Angew. Chem. **2015**, 127, 12212–12216

- [1] Typing the name of these drug and drug candidates into the NCGC Pharmaceutical Collection database provides compound structure, bioactivity, a full list of literature, and access to ongoing clinical trials, applications, and usage.
- [2] a) M. Filizola, L. A. Devi, Nature 2012, 485, 314–317; b) A. Manglik, A. C. Kruse, T. S. Kobilka, F. S. Thian, J. M. Mathiesen, R. K. Sunahara, L. Pardo, W. I. Weis, B. K. Kobilka, S. Granier, Nature 2012, 485, 321–326; c) H. X. Wu, D. Wacker, M. Mileni, V. Katritch, G. W. Han, E. Vardy, W. Liu, A. A. Thompson, X. P. Huang, F. I. Carroll, S. W. Mascarella, R. B. Westkaemper, P. D. Mosier, B. L. Roth, V. Cherezov, R. C. Stevens, Nature 2012, 485, 327–332; d) S. Granier, A. Manglik, A. C. Kruse, T. S. Kobilka, F. S. Thian, W. I. Weis, B. K. Kobilka, Nature 2012, 485, 400–404.
- [3] Synthesis of ketorfanol from the degradation of heavily functionalized natural product starting materials: a) A. Manmade, H. C. Dalzell, J. F. Howes, R. K. Razdan, J. Med. Chem. 1981, 24, 1437–1440; b) Y. Ida, T. Nemoto, S. Hirayama, H. Fujii, Y. Osa, M. Imai, T. Nakamura, T. Kanemasa, A. Kato, H. Nagase, Bioorg. Med. Chem. 2012, 20, 949–961.
- [4] Reviews and leading references on the syntheses of morphine and related natural products: a) J. W. Reed, T. Hudlicky, Acc. Chem. Res. 2015, 48, 674-687; b) U. Rinner, T. Hudlicky, Top. Curr. Chem. 2012, 300, 33-66; c) N. Chida, Top. Curr. Chem. 2011, 299, 1-28; d) M. Tissot, R. J. Phipps, C. Lucas, R. M. Leon, R. D. M. Pace, T. Ngouansavanh, M. J. Gaunt, Angew. Chem. Int. Ed. 2014, 53, 13498-13501; Angew. Chem. 2014, 126, 13716-13719; e) M. Ichiki, H. Tanimoto, S. Miwa, R. Saito, T. Sato, N. Chida, Chem. Eur. J. 2013, 19, 264-269; f) J. Li, G.-L. Liu, X.-H. Zhao, J.-Y. Du, H. Qu, W.-D. Chu, M. Ding, C.-Y. Jin, M.-X. Wei, C.-A. Fan, Chem. Asian J. 2013, 8, 1105-1109; g) V. Varghese, T. Hudlicky, Angew. Chem. Int. Ed. 2014, 53, 4355-4358; Angew. Chem. 2014, 126, 4444-4447; h) A. Kimishima, H. Umihara, A. Mizoguchi, S. Yokoshima, T. Fukuyama, Org. Lett. 2014, 16, 6244-6247.
- [5] a) D. A. Colby, R. G. Bergman, J. A. Ellman, J. Am. Chem. Soc. 2008, 130, 3645 3651; b) S. Duttwyler, C. Lu, A. L. Rheingold, R. G. Bergman, J. A. Ellman, J. Am. Chem. Soc. 2012, 134, 4064 4067; c) S. Duttwyler, S. Chen, M. K. Takase, K. B. Wiberg, R. G. Bergman, J. A. Ellman, Science 2013, 339, 678 682; d) M. A. Ischay, M. K. Takase, R. G. Bergman, J. A. Ellman, J. Am. Chem. Soc. 2013, 135, 2478 2481; e) S. Duttwyler, S. Chen, C. Lu, B. Q. Mercado, R. G. Bergman, J. A. Ellman, Angew. Chem. Int. Ed. 2014, 53, 3877 3880; Angew. Chem. 2014, 126, 3958 3961; < lit f > T. Mesganaw, J. A. Ellman, Org. Process Res. Dev. 2014, 18, 1097 1104; g) T. Mesganaw, J. A. Ellman, Org. Process Res. Dev. 2014, 18, 1105 1109.
- [6] Intramolecular Friedel Crafts alkylation has been used to form this ring system, but not in the presence of the acid-labile allylic alchol functionality. R. Grewe, A. Mondon, *Chem. Ber.* 1948, 81, 279–286.

- [7] A discussion on redox economy in synthesis: N. Z. Burns, P. S. Baran, R. W. Hoffmann, *Angew. Chem. Int. Ed.* 2009, 48, 2854–2867; *Angew. Chem.* 2009, 121, 2896–2910.
- [8] A recent review on asymmetric electrocyclization reactions: S. Thompson, A. G. Coyne, P. C. Knipe, M. D. Smith, *Chem. Soc. Rev.* 2011, 40, 4217–4231.
- [9] a) R. P. Hsung, L. L. Wei, H. M. Sklenicka, C. J. Douglas, M. J. McLaughlin, J. A. Mulder, L. J. Yao, Org. Lett. 1999, 1, 509-512;
 b) K. Tanaka, S. Katsumura, J. Am. Chem. Soc. 2002, 124, 9660-9661;
 c) H. M. Sklenicka, R. P. Hsung, M. J. McLaughlin, L. I. Wei, A. I. Gerasyuto, W. B. Brennessel, J. Am. Chem. Soc. 2002, 124, 10435-10442;
 d) N. Sydorenko, R. P. Hsung, E. L. Vera, Org. Lett. 2006, 8, 2611-2614;
 e) T. Kobayashi, K. Takeuchi, J. Miwa, H. Tsuchikawa, S. Katsumura, Chem. Commun. 2009, 3363-3365;
 f) T. Sakaguchi, S. Kobayashi, S. Katsumura, Org. Biomol. Chem. 2011, 9, 257-264.
- [10] Recent reports of remote stereocontrol of triene electrocyclization: a) Z.-X. Ma, A. Patel, K. N. Houk, R. P. Hsung, *Org. Lett.* 2015, 17, 2138–2141; b) A. Patel, G. A. Barcan, O. Kwon, K. N. Houk, J. Am. Chem. Soc. 2013, 135, 4878–4883; c) G. A. Barcan, A. Patel, K. N. Houk, O. Kwon, *Org. Lett.* 2012, 14, 5388–5391.
- [11] D. Q. Xu, G. A. Crispino, K. B. Sharpless, J. Am. Chem. Soc. 1992, 114, 7570 – 7571.
- [12] Synthetic details, characterization, and molecular structures obtained by X-ray structural analysis of compounds are described in the Supporting Information. CCDC 1404582 (11) and 1404583 (1·HCl) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.
- [13] K. A. Davies, R. C. Abel, J. E. Wulff, J. Org. Chem. 2009, 74, 3997–4000.
- [14] C. H. Larsen, K. W. Anderson, R. E. Tundel, S. L. Buchwald, Synlett 2006, 2941–2946.
- [15] A. Fürst, P. A. Plattner, Helv. Chim. Acta 1949, 32, 275-283.
- [16] Computations of the electrocyclic reaction of carbatriene analogue of **E** predict a similar level of torquoselectivity.



[17] A leading reference: J. C. Lewis, A. M. Berman, R. G. Bergman, J. A. Ellman, J. Am. Chem. Soc. 2008, 130, 2493–2500.

Received: June 17, 2015

Published online: August 17, 2015